A LEPTOSIDIN GLYCOSIDE FROM LEAVES OF CYPERUS SCARIOSUS

S. K. BHATT, V. K. SAXENA and K. V. SINGH*

Department of Chemistry and *Department of Botany, University of Saugar, Sagar, M.P. 470003, India

(Revised received 6 February 1981)

Key Word Index—Cyperus scariosus; Cyperaceae; aurone; leptosidin 6-glucosylrhamnoside.

Abstract—Phytochemical examination of the leaves of *Cyperus scariosus* resulted in the isolation and identification of a new glycoside, leptosidin $6-O-\beta$ -D-glucopyranosyl- $O-\alpha-1$ -rhamnopyranoside.

Phytochemical examination of this plant was undertaken because little work has been done on the leaves [1, 2]. A phenolic glycoside, mp 232-234°, was isolated, which, upon hydrolysis with 7% H₂SO₄ in EtOH yielded an aglycone, glucose and rhamnose. The aglycone crystallized from petrol-Me₂CO as orange-yellow crystals, mp 253-255°. (Found: C, 63.40; H 3.8 Calc. for $C_{16}H_{12}O_6$, C, 63.5, H 4.0%.) It changed colour from yellow to orange-red and finally to red with alkali and produced a bright red colour with conc. H₂SO₄, indicating the presence of an aurone. The aglycone showed the presence of one methoxyl group (Zeisel) and also afforded a triacetate on acetylation with (Ac₂O/pyridine). Methylation of the aglycone gave a trimethyl ether with mp 155-156° which on oxidation with KMnO₄ yielded veratric acid, mp 170°, which indicates the presence of free hydroxyls at 3' and 4'. The IR spectrum showed prominent peaks at 3336 (OH), 2901 (-C-H) stretch) 1681 (-C-C-O-stretch) 1605. 1563, 1441 (ring system) and 1283 cm^{-1} (-C-O-C vibration). The UV shift of the aglycone with NaOAc and AlCl₃ was exactly similar to that of leptosidin [3, 4] while the glycoside did not show a UV shift for a free 6-hydroxyl [5]. Therefore the aglycone was identified as leptosidin, which was confirmed by co-TLC and mps with an authentic sample [6]. Periodate oxidation of the glycoside consumed 3.2 mol with the liberation of 1.3 mol of formic acid/mol of the glycoside indicating the presence of a disaccharide with both sugars in the pyranose form. The glycoside on partial hydrolysis (with 2% H₂SO₄) gave glucose, suggesting that it was the terminal sugar and that rhamnose is linked with the aglycone at position 6. The permethylated glycoside [7] on acid hydrolysis vielded 2,3-di-O-methyl-L-rhamnose (by co-TLC with authentic) and 2,3,4,6-tetra-O-methylglucose (mps and co-TLC with an authentic sample) indicating that the glucopyranose unit is joined to the rhamnopyranose unit by a $1 \rightarrow 4$ -linkage. The enzymatic hydrolysis [8] of the glycoside showed a β -linkage between the two sugars and an α -linkage between the aglycone and the rhamnose.

Therefore the structure of new glycoside was assigned as leptosidin 6-O- β -D-glucopyranosyl-O- α -L-rhamnopyranoside.

EXPERIMENTAL

Air-dried leaves (2 kg) of Cyperus scariosus were obtained from Saugar region and identified through the courtesy of the Botany Department, University of Saugar. The leaves were extrd with 95% EtOH under reflux for 20 days. The extract (2.51.) was concd under red. pres. to 125 ml and stored in a refrigerator whereupon a syrupy green deposit was formed. The filtrate was poured into distilled water (500 ml) with continuous stirring. The water-soluble portion was extrd with EtOAc which gave the glycoside. It was purified on a column of si gel and cryst. as yellow needles from petrol–Me₂CO, mp 232–234°. The homogeneity of the glycoside was checked by PC in n-BuOH–HOAc–H₂O (yield 0.031%).

Acknowledgements—The authors thank Director, Glaxo Co., Bombay for spectral analysis and also one of the authors (S.K.B.) is thankful to UGC for financial help.

REFERENCES

- Singh, K. V. and Agrawal, S. C., Indian Drugs Pharm. Industry (in press).
- Singh, K. V. and Pathak, R. K. (1979) Indian Drugs Pharm. Industry 14, 25.
- Geissman, T. A. and Heaton, C. D. (1943) J. Am. Chem. Soc. 65, 677.
- Geissman, T. A. and Heaton, C. D. (1944) J. Am. Chem. Soc. 66, 486.
- Harborne, J. B., Mabry, T. J. and Mabry, H. (1974) The Flavonoids, p. 46. Chapman & Hall, London.
- 6. Nigam, S. S. and Saxena, V. K. (1975) Planta Med. 27, 98.
- 7. Khun, R., Trischmann, H. and Low, I. (1955) Angew. Chem. 67, 32.
- 8. Mann, F. G. and Saunders, B. C. (1936) Practical Organic Chemistry, p. 365. Longman, New York.